between $O(2)$ and $H(1)$ and a H bond type linkage (albeit a strong one) between $O(3)$ and $H(1)$.

The very strong interactions of this arrangement are partially relieved in 2 ways. As in the KHMal, KHClMal, and maleio acid structures the internal angles of the ring system are strained considerably. The two angles $O(2)-C(1)-C(2)$ and $O(3)-C(4)-$ C(3) are not significantly different from each other and average 119[°] 36' \pm 42'. Similarly the angles C(1)–C(2)–C(3) and C(2)– C(3)-C(4) are insignificantly different and average 130° 24' \pm 24'. These angles are however significantly different from their unstrained counterparts of 121° 18' and 121° 30', respectively.²⁴ A situation closely paralleling this has been found in the 3 structures mentioned above.

The second manner in which the strain is relieved is by rotation of the C02H groups around their C-C bonds. The atoms of the C spine here are all within 0.007 \AA of the plane with equation $-0.815x + 0.4940y - 0.3027z - 6.0592 = 0.$ $(x^2 = 3.6)$. The $O(1)$, $O(2)$, $O(3)$, and $O(4)$ atoms on the other hand are $0.163(5)$, $-0.156(4)$, $-0.133(4)$, and $0.126(4)$ Å from this plane. These figures imply that the torsion angle about $C(1)-C(2)$ for $CO₂H$ 1 is 8° 23' and that about $C(3)-C(4)$ for $CO₂H$ 2 is 6° 46' the directions of twist being such as to put $O(2)$ and $O(3)$

(24) S. F. Darlow, *Acta Crystallogr.,* 14, 1257 (1961).

on the same side of the plane of the carbons. A view of the ion showing this conformation is included in Figure 5.

A similar situation was found in KHClMal but not in KHMal or maleic acid. At present there seems little reason for $O(2)$ and $O(3)$ being on the same side of the C spine but is is intended to investigate the structures of disodium maleate and maleic acid in the hope of shedding some light on this question, or at least partially delineate its occurrence.

Acknowledgments.—We thank the Schering Corporation for the sample of dl -brompheniramine maleate used in this analysis. Doctors A. F. Casy and R. R. Ison of the Department of Pharmacy of this University originally suggested this project to us and have continued to be a source of information regarding it. One of us (G. W.) would like to thank Dr. Ison for a copy of his Doctoral Dissertation. Messrs. L. P. Simonsen and R. J. Ulrich worked on this project as senior undergraduates and we thank them for their help. Financial support from the Medical Research Council of Canada is gratefully acknowledged. The crystallographic computer programs of Dr. F. R. Ahmed, National Research Council of Canada have been invaluable in this work.

Steroidal Androgen Biosynthesis Inhibitors

G. E. ARTH,* A. A. PATCHETT, T. JEFOPOULUS, R. L. BUGIANESI, L. H. PETERSON,

Synthetic Chemical Department, Merck Sharp & Dahme Research Laboratories, Rahway, New Jersey 07065

E. A. HAM, F. A. KUEHL, JR., AND N. G. BRINK

Merck Institute for Therapeutic Research, Merck Sharp & Dohme Research Laboratories, Rahway, New Jersey 07065

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A screening procedure using a rat testicular microsomal preparation with $[21-14C]17\alpha$ -hydroxyprogesterone as substrate was used to search for inhibitors of 17,20-lyase. A series of 17 β -acylaminoandrost-4-en-3-ones and derivatives has been prepared and their syntheses have been described; they are shown to be androgen synthesis inhibitors *in vitro* and *in vivo.* These steroidal androgen synthesis inhibitors are demonstrated to be more specific in their action than the nonsteroidal inhibitors previously known. A mechanism for the action of these compounds is postulated.

Selective control of androgen biosynthesis is of potential interest in the treatment of benign prostatic hypertrophy, hirsutism, acne, and androgen-dependent tumors. Nonsteroidal synthesis inhibitors are known¹ which display varying degrees of selectivity in respect to enzymes involved in the adrenal and testicular synthesis of steroids including androgens.²

The major pathways^{2,3} which have been established for testicular androgen biosynthesis involve hydroxylation of progesterone and pregnenolone to the 17α hydroxyl derivatives and cleavage by 17,20-lyase(s) to yield androst-4-ene-3,17-dione and 3β -hydroxyandrost-5-en-17-one, respectively. Testosterone arises by reduction of the former at C-17 and in the latter by the combined action of $3(\text{or } 17)\beta$ -hydroxy steroid: NAD (P) oxidoreductase, 1.1.1.51, and 3-keto steroid Δ^5 , Δ^4 isomerase, 5.3.3.1. The possibility of direct formation of testosterone from progesterone has also been sug-

gested but not conclusively established.⁴ Of several possibilities we chose to search for inhibitors of the 17,20-lyase step in androgen synthesis because of its key role in the conversion of C-21 to C-17 steroids and in the hope that the enzyme might show considerable structural specificity with respect to inhibitors. Nonsteroidal inhibitors were not excluded from consideration but steroidal inhibitors were judged to have the best chance to show a selective action.

Assay Method.—An *in vitro* screening procedure was used which involved measurement of [¹⁴C]acetate formed in the side-chain cleavage of $[21^{-14}C]17\alpha$ -hydroxyprogesterone⁵ by a rat testicular microsomal preparation. (See footnote *a,* Table I for the protocol used.)

Chemistry.—The first significant observation was that testosterone acetate but not testosterone was a potent inhibitor of the 17,20-lyase, and prompted a

⁽¹⁾ R. Neher and F. W. Kahnt, *Experientia,* 21, 310 (1965).

⁽²⁾ For an excellent review of this general subject see R. Gaunt, B. G. Steinet_', and J. J. Chart, *Clin. Pharmacol. Ther.*, 9, 657 (1968).
(3) J. H. Richards and J. B. Hendrickson, "The Biosynthesis of Steroids,

Terpenes and Acetogenins," W. A. Benjamin, Inc., New York, N. Y., 1964, pp 351-357.

^{(4) (}a) R. I. Dorfman, E. Forchielli, and M. Gut, *Recent Progr. Horm. Res.,* 19, 14 (1963); (b) R. I. Dorfman and F. Ungar, "Metabolism of Steroid Hormones," Academic Press, New York, N. Y., 1965, p 1127; (c) M. A. Drosdowky, E. Forchielli, and R. I. Dorfman, *J. Eur. Steroides,* 2, 515 (1967).

⁽⁵⁾ Kindly supplied to us by Dr. Robert E. Erickson, formerly of these laboratories.

synthetic program to make related but metabolically stable analogs of testosterone acetate. This was approached by making amide, urea, guanidino, and carbamate substitutions for the 17β -OAe group.

Scheme I summarizes the procedures used to make

many of the acyl derivatives and conversion products of 176-aminoandrost-5-en-36-ol.⁶ A Beckamn rearrangement of the optionally substituted 3β -acetoxypregn-5-en-20-one 20-oxime afforded the corresponding 17/3-acetamidoandrost-5-en-3/3-yl acetate. This process is one first described by Schmidt-Thome.⁶¹' In those cases in which substitution other than 17β -CH3COXH was desired, strong alkaline hydrolysis was employed to afford the 17β -amino derivative which was converted to other 17β -acyl or -ureido functions by standard means. Oppenauer oxidations were used to convert the 3β -hydroxy-5-ene functionality to 3-keto-4ene derivatives and DDQ oxidations afforded the 3 keto-l,4-diene structures from the latter. Most of these conversions proceeded without difficulty. However, an exception was the Oppenauer oxidation of 17β ureidoandrost-5-en-3 β -ol (9) which could only be accomplished in good yield after X³ -acetylation of the

(6) (a) G. Ehrhart , H. Ruschig, and W. Aumllller, *Angew. Chem.,* 52, 363 (1939); (b) J. Schmidt-Thorn^ , *Chem. Ber.,* 88, 895 (1955).

17/3-ureido function (see Experimental Section for details).

Because of the unanticipated difficulty encountered in the Oppenauer oxidation of 17β -ureidoandrost-5-en- 3β -ol to 7 a second route to the latter was employed using 17β -aminoandrost-5-en-3-one ethylene cyclic acetal⁷ which afforded an identical final product. This intermediate also served as a convenient starting material for other derivatives, notably 6.

The procedure of Joska and Sôrm^s was the method of choice for the synthesis of 4 from which a derivative 22 was prepared.

The preparation of N-methyl-17 β -acetamidoandrost-4-en-3-one (19) was accomplished by X-methylation of 17β -acetamidoandrost-5-en-3 β -yl acetate followed by partial hydrolysis and oxidation. The isomeric 17α methyl-17 β -acetamidoandrost-4-en-3-one (18) was prepared from 17α -methylpregnenolone acetate⁹ by the standard procedure.

Synthesis of 17β -acetamidoestr-4-en-3-one (20) proceeded from 3-methoxyestra-l,3,5(10)-trien-17-one oxime¹⁰ which was reduced with Na in EtOH⁹ to the 17 β amino compound. Standard Birch reduction conditions afforded the dihydro derivative which, without purification, was acetylated and then treated with acid affording 20. To further delineate the relation of structure to activity two 17α -acylamido derivatives 10 and 25 were prepared through a common intermediate 17α -aminoandrost-5-en- 3β -ol¹¹ by methods described in the Experimental Section.

Relationship of Lyase Inhibition with Structure. It is evident from an examination of the data presented in Table I that high inhibition was associated with androst-4-en-3-ones bearing substituents at C-17 β closely related to $CH₃CO₂$ in size and polarity. Thus **3** (17 β -OCONH₂), 4 (17 β -NHCOH), 5 (17 β -NHCOCH₃), and 7 (17 β -NHCONH₂) were all highly active. Larger groups at C-17 were associated with decreased activity as was epimerization at C-17 or by 17α substitution as illustrated by 18 and 23. The 17β -guanidino analog, (12) of androst-5-ene-3 β ,17 β -diol was virtually inactive. Considerable tolerance, however, was permitted in respect to changes around the A ring of the inhibitors. In this *in vitro* system virtual equivalence to the corresponding 4-en-3-one analog was found with 9 and 14 $(5\text{-}en-3\beta\text{-}ol)$, 8 and 22 $(1,4\text{-}dien-3\text{-}one)$, and 13 $(4\text{-}en 3\beta$ -ol) and 15 (4,5- α -dihydro-3-one). Methylation at 6α (16) was consistent with good activity but 16α methylation (17) was not. In connection with the latter it is of interest that 16α -methylation also markedly decreases androgenic activity. A 19-nor variant (20) was active but this change did not increase potency. An aromatic A ring derivative (28) was inactive.

Specificity of Inhibition.—Compounds 4, 5, 7, 9, 13, and 24 did not inhibit the *in vitro* conversion of chlolesterol to pregnenolone by an acetone preparation from bovine corpora lutea, indicating that these compounds

⁽⁷⁾ Von A. Schubert, K, Zepter, II. Cireiner, and E. Watzke , •/. *Pnikt. Chem.,* 26, 324 (1964).

⁽⁸⁾ J. Josk a and F. Sorm, *Colled. Czech. Chem. Commun.,* 21 , 761 (1956); *Chem. Listy*, 49, 1687 (1955); *Chem. Abstr.*, 56, 5715.

⁽⁹⁾ Pl. A. Plattner. H. Heusser, and P. Th. Herzig, *Helv. Chim. Acta.*, 32, 270 (1949).

⁽¹⁰⁾ B. M. Regan and F. N. Hayes, *J. Amer. Chem. Soc.*, **78**, 639 (1956).

^{(11) (}a) J. W. Cole, *Chem. Abstr.*, 62, 5319 (1965); (b) C. H. Robinson and C. Ermann, *Steroids*, 6, 509 (1965).

TABLE I ACTIVITIES OF SOME OF THE 17,20-LYASE INHIBITORS

" The incubation mixt contained 0.60 mg of testicular microsomal protein prepd from mature Holtzman rats, 5 mg (50,000 dpm) of [21-¹⁴C] 17a-hydroxy progesterone, 0.74 mg of NADPH, and 20 Ml of i-PrOH in 2.0 ml of 0.02 *M,* pH 7.0, Tris buffer. The incubation was allowed to proceed at 37° for 30 min, and the [2-¹⁴C]AcOH which was formed was sepd from the mixt with the use of a 1.5-ml column of Dowex 1-X8 resin of 200-400 mesh in the incubation buffer. $H_2SO_4(3.0 \text{ ml}, 0.5 N)$ was used to elute the $[2^{-14}C]$ AcOH. Control incubations produced 30% cleavage. The screening level for inhibitors was set at 0.5 μ g/ml. Compds producing <10% inhibition at this concn were considered to be inact. The percentage inhibition is reported to the nearest 5% and is reproducible to ±2.5%. ^b This communication, see Experimental Section. \cdot See ref 8. ^d M. F. Murray, U. S. Patent 2,707,189 (1955); *Chem. Abstr.*, 50, 4246/ (1956). *'* By Oppenauer oxidn of 16a-methyl-17/3-acetamidoandrost-5-en-30-ol, described by P. DeRuggieri, C. Ferrari, and C. Gandolfi, *Gazz. Chim. Ital,* 91, 655 (1961). / O. El-Tayeb, S. G. Knight, and C. J. Sih, *Biochem. Biophys. Acta,* 93, 411 (1964). « A. A. Patchett, F. Hoffman, F. F. Giarrusso, H. Schwam, and G. E. Arth, *J. Org. Chem.,* 27, 3822 (1962). * We are indebted to Dr. Emil Schlittler of Ciba Pharmaceutical Company, Summit, N. J., for generous supplies of this compd. * Confirmation of the findings of D. C. Sharma, E. Forchielli, and R. I. Tamaoki, *Biochim. Biophys. Acta.,* 105, 516 (1965), and refs cited. ' See ref 7.

are not general lyase inhibitors.¹² Further, 5, 7, and 9 did not inhibit the *in vitro* conversion of corticosterone to 18-hydroxycorticosterone by crude adrenal mitochondria,¹² whereas in the same assay 26 and 27 did significantly $(>20\%$ at 0.01 μ g/ml of incubate) inhibit this conversion. It thus appears that the steroidal lyase inhibitors herein described are more specific in their action than the nonsteroidal inhibitors 26 and 27. Indirect evidence indicating specificity, particularly that 21- and 11β -hydroxylation are not inhibited, was obtained by feeding 8 to male rats at 500 mg/kg of diet for 6 weeks.¹³ Adrenal weights of the treated animals were not significantly different from those of controls. Inhibition of 21- and/or 11 β -hydroxylases would be expected to result in adrenal hypertrophy.

Testes of the rats treated as above showed a $75-90\%$ decrease in testosterone levels as compared with controls.¹⁴ This is considered to be a qualitative *in vivo* correlation with predictions based on the *in vitro*

⁽¹²⁾ We are indebted to Thomas Gutwein for the numerous 17,20- and 20,22-lyase inhibition assays herein reported. We also thank James Fridie for the 18-hydroxylation observations. The method used for measuring the cholesterol to pregnenolone conversion is that of S. Ichii, E. Forichelli, and R. I. Dorfman, *Steroids,* 2, 631 (1963), and for 18-hydroxylation that of D. C. Sharma and R. I. Dorfman, *Biochemistry,* 8, 1795 (1966).

⁽¹³⁾ D. J. Patanelli, C. Berman, R. Primka, and S. L. Steelman, of these laboratories, unpublished results.

⁽¹⁴⁾ The testosterone content of rat testes was determined as follows. Approximately 20 rat testes were homogenized and spiked with tritrated testosterone (5000 eps). Acetone (10 vol) was added to ppt protein which was sepd by centrifugation. The supernatant was reduced to an aq residue in a rotary evaporator. H₂O was added to a total of 10 ml, and the mixt was extd $(3 \times 15 \text{ m})$ with EtOAc. The EtOAc ext was evapd to dryness in a stream of N_2 . The residue was partitioned between hexane and a 90% MeOH-10% H₂O mixt. The hexane ext containing nonpolar lipids was M eOH-10% H₂O mixt. The hexane ext containing nonpolar lipids was discarded. The aq MeOH phase was coned to an aq residue in a stream of N₂; 2 ml of H₂O was added and the mixt was extd with EtOAc $(3 \times 3$ ml), the aq layer containing water-sol impurities being discarded. This ext was adsorbed on a silica gel tic and developed with 50:50 EtOAc-cyclohexane. The testosterone spot was eluted with EtOAc and evapd. An aliquot of this eluate was applied to a 1.7% S.E. 30 gle column whence the mass of the testosterone was detn. Another aliquot was counted (to detn yield through the process). The androstenone (A) and dehydroisoandrosterone (DHA) spots were also eluted from the tic plate and their masses detn by glc. Their mass detns were corr using the yield figure from the testosterone isolation. mass det as were corrusing the yield lighter from the testosteroide isolation.
Since the A and DHA content of testes of treated rats were also substantially reduced as compared with controls it is clear that the drug is not merely changing the ratios of the testicular androgens.

results. None of the acylaminoandrostenones is antiandrogenic.¹⁵

Nature of the Inhibition.—The structures of these inhibitors suggest that they are reversible antagonists which bear a homologous relationship to the lyase substrate 17α -hydroxyprogesterone. This lyase reaction requires O_2 and NADPH and the ensuing reaction bears an apparent formal resemblance to the Baeyer-Villiger oxidation. The expected intermediate based on this analogy, 17α -hydroxy-17 β -acetoxyandrost-4-en-3-one, would not be expected to be stable, breaking down to androstenedione and AcOH. This postulated mechanism, which was first proposed by Xakano, *el* a/.,¹⁶ was based on the observations of Fonken, *et al.*,^{*i*} Rahim and Sih,¹⁸ and Nakano, *et al.*,¹⁹ that progesterone is aerobically converted directly to testosterone acetate and thence to testosterone by *Cladosporium resinae.* We hypothesize that inhibitors of the *in vitro* rat lyase system such as testosterone acetate and the other analogs described in this paper resemble an intermediate or transition state on the enzyme at which a separation of the C-17,20 carbon atoms occurs. The inhibitory compounds, however, lack a 17α -OH group and therefore there is no pathway to products.

Experimental Section²⁰

General Methods. I. Oximes.^{-The} general method used for the prepns of compds A (Scheme I) is that described in ref 6 (see footnote *b,* Table I). By this method the following new compds were obtained: **6-methylpregnenolone acetate 20** oxime (29) (from 6-methylpregnenolone acetate²¹), mp 195-200° dec, from MeOH, $[\alpha]_D - 70^{\circ}$, anal. (C₂₄H₃₆NO₃) C, H, N; **17a-methylpregnenolone acetate 20-oxime (30)** (from 17amethylpregnenolone acetate⁹), mp 194-196°, from MeOH, $[\alpha]$ D -48.3° , anal. (C₂₄H₃₇NO₃) C, H, N.

II. Beckman Rearrangement.—The general method used for the prepn of compds B (Scheme I) is that described in ref 6 (see footnote *b,* Table I). By this method the following new compds were obtained: 6-methyl-17 β -acetamidoandrost-5-en-3 β -yl ace**tate** (31) (from 29), mp 204-207°, from Me₂CO, $[\alpha]_D$ -113°, anal. (C₂₄H₃₇NO₃) C, H, N; 17a-methyl-17³-acetamidoandrost-**5-en-30-yl acetate (32)** (from 30), mp 194-196°, from MeOH, *anal.* ($C_{24}H_{36}NO_3$) C, H, N.

III. O-Deacylation.—The general method vised for the prepn of compds C $(A' = H)$ (Scheme I) is that described in footnote d, Table I. By this method the following new compds were obtained: 6 -methyl-17 β -acetamidoandrost-5-en-3 β -ol (33) (from 31), mp 255-258°, from MeOH, $[\alpha]$ D -116°, *anal.* (C₂₂H₃₅NO₂) C, H, N; 16α -methyl-17 β -acetamidoandrost-5-en-3 β -ol (34) (footnote ϵ , Table I); 17α -methyl-17 β -acetamidoanerost-5-en-3 β -ol (35) (from 32), mp 284-286°, from MeOH, anal. (C₂₂H₃₅NO₂) C, H, N. 17 β -. V-Methylacetamidoandost-5-en-3 β -ol was not characterized but was oxidized to the 3-keto- Δ^4 deriv as described below.

IV. Oppenauer Oxidn.—The general method for the prepn of compds I) (Scheme I) is that given in footnote *d,* Table I.

(17) G. S. Fonken, H. C. Murray, and L. M. Reinecke, *J. Amer. Chem. Soc,* 82, 5507 (1960).

(19) H. Nakano, H. Sato, and B. Tamaoki, *Steroids,* **12,** 291 (1968).

(21) Purchased from com sources.

By this method the following compds were prepd: 6α -methyl-**17/3-acetamidoandrost-4-en-3-one (16)** (from **33),** mp 298-301°, from EtOAc, $[\alpha]_D$ +15°, *anal.* (C₂₂H₃₃NO₂) C, H, N; 16 α methyl-17 β -acetamidoandrost-4-en-3-one (17) (from 34), mp 311-315°, from MeOH, $[\alpha]_D + 23^\circ$, *anal.* (C₂₂H₃₃NO₂) C, H, N; 17α -methyl-17 β -acetamidoandrost-4-en-3-one (18) (from 35), mp 232-234°, from MeOH, $[\alpha]_D$ +66.6°, anal. $(C_{22}H_{33}NO_2)$ C, H, N; 17α -acetamidoandrost-4-en-3-one (25) (from 17α acetamidoandrost-5-en- β -ol¹¹), mp 252-254°, from EtOAc, $[\alpha]_D$ +139.5°, *anal.* (C₂₁H₃₁NO₂) C, H, N; 17 β -(N³**-acetureido)androst-4-en-3-one (11)** (from 39), mp 270-272°, from MeOH, *anal.* (C22H32N203) C, H, N; **17/3-A^r -methyIacetamidoandrost-4-en-3-one (19)** (from 17/3-A^r -methylacetamidoandrost-5-en-3 β -ol), mp 186-188°, from MeOH, [a] $p +61.8$ °, anal. (C₂₂H₃₃- $NO₂)$ C, H, N.

V. **l,4-Dien-3-ones.**—The general procedure is illustrated by the conversion of 17β -acetamidoandrost-4-en-3-one (5) to 17β -acetamidoandrosta-1,4-dien-3-one (21) as follows. A mixt of 5 (4.5 g), DDQ, and dioxane (90 ml) was heated at reflux under N_2 for 3 hr. Dissoln of all of the reactants occurred and the color changed to deep red. EtOAc (200 ml) was added to the cooled soln and the insol hydroquinone was sepd by filtration and washed once with EtOAc (50 ml). The filtrate was washed with aq 2.5 N NaOH which removed most of the color and then with H_2O to neutrality. The org layer was dried (Na_2SO_4) and evapd under reduced pressure leaving crude dienone which was purified by chromatog on Al₂O₃. The product 21, 4.1 g, eluted with a 7:3 CHCl₃-Et₂O mixt, was recrystd from EtOAc-CHCl₃, mp 270-272° (cf. footnote f, Table I).

Using the above procedure, 4 was converted to 17 β -formamido**androsta-l,4-dien-3-one (22),** mp 227-228°, from EtOAc- $CHCl₃, [\alpha]_{\text{D}}$ +53°. *Anal.* $(C_{20}H_{27}NO_2)$ C, H, N. Also 11 was converted to 17β -(N³-acetureido)androsta-1,4-dien-3-one **(40),** mp 264-265°, from MeOH, [a]D +92.8°. *Anal.* (C22- $H_{30}N_2O_2$) C, H, N.

17 β **-Aminoandrost-5-en-3** β **-ol·HCl (36)** was prepd by a slight modification of the method of Schmidt-Thome^{6b} which, however, gives a superior product. A soln of 45 g of NaOH in 150 ml of H_2O was added to a soln of 5 g of 17 β -acetamidoandrost-5-en- 3β -yl acetate²⁰ in 350 ml of EtOH. A clear soln resulted. The reaction mixt was heated at 180° in a stirred autoclave for 8 hr under N_2 . The contents of the cooled autoclave were filtered through Supercel and the clear, nearly colorless filtrate was evapd under reduced pressure until a thick syrup remained. $H₂O$ (500 ml) was added, and the mixt was extd with CHCl₃ $(2 \times 300 \text{ ml})$. The combined CHCl₃ exts were washed once with H₂O (50 ml) which was discarded. To the CHCl₃ was added with $H₂O$ (50 ml) which was discarded. excess 2.5 A" aq HC1 to ppt the amine-HC1, which was then sepd by filtration, washed once with Me₂CO, and dried in a stream of air, yield 40 g, mp (dec) *ca*. 300°.²² This colorless product was of excellent quality and was used in subsequent reactions without further purification.

17/3-A^r -Methylacetamidoandrost-5-en-3/3-yl Acetate (37).— To a soln of 1.7 g of 17 β -acetamidoandrost-5-en-3 β -yl acetate in 35 ml of dry \rm{DMF} was added 1.2 g of a 54.7% dispersion of \rm{NaH} in mineral oil, and the resulting mixt was stirred at room temp for 1 hr. The reaction mixt was then cooled to 0° and to this was added in a dropwise manner a soln of freshly distd Mel (20 ml) in 17 ml of dry DMF. The reaction mixt was then permitted to warm to room temp and stirred overnight. After cooling to 0°, H² 0 was cautiously added to destroy the remaining NaH, followed by sufficient H_2O to ppt the product which was extd into $CHCl₃$ (100 ml). The CHCl₃ ext was thoroughly washed with $H₂O$ (5 \times 20 ml), dried, and evapd at reduced pressure, finally at 0.1 mm, to remove the remaining DMF. The residue was adsorbed on a 40:1 alumina column from a CeHs soln. The product 37, was eluted with increasing conens of CHCl₃ in C_6H_6 and crystd from MeOH, yield 1.6 g , mp 199°. Anal. (C₂₄- $H_{37}NO_3$) C, H.²³

17/3-Ureidoandrost-5-en-3/3-ol (9).—To a suspension of 5 g of 36 and 50 ml of EtOH was added 7.5 g of KOCN. The mixt was heated for 2-3 min on the steam bath and then treated with 7.5 ml of 2.5 *N* HC1. The mixt was then heated at reflux in the steam bath for 15 min and filtered while hot. H_2O was added to the cooled filtrate until crystn was complete. The product 9 was sepd by filtration and washed well with H₂O. It had mp 320°

⁽¹⁵⁾ A definition of this term is implicit in the method used in its detn. See S. L. Steelman, J. R. Brooks, E. R. Morgan, and D. J. Patanelli, *Steroids,* **14,** 449 (1969).

⁽¹⁶⁾ H. Nakano, H. Ivano, H. Sato, M. Siiikita, and B. Tamaoki, *Biochim. Biophys. Acta,* **137,** 335 (1967).

⁽¹⁸⁾ M. A. Rakim and C. J. Sih, *J. Biol. Chem.,* **241,** 3615 (1966).

⁽²⁰⁾ Melting points are uncorr and were detd on a Kofler hot stage. Rotations were run in CHCl₃ (c 1) unless otherwise noted and are expressed as $[\alpha]^{25}$ D. Uv spectra were determined in EtOH and all 3-keto- Δ^4 and 3keto- $\Delta^{1/4}$ compounds absorbed at 240-242 mu and showed ϵ_{max} ca. 16,000. All of the new compds showed ir spectra consistent with the assigned struc-Nmr spectra were run on selected compds and were consistent with the findings of Robinson and Ermann.^{10b} Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values.

⁽²²⁾ Swiss Patent 187,936; *Chem. Abstr.,* **31,** 5109 (1937).

⁽²³⁾ Unpublished procedure of Dr. N. G. Steinberg of these laboratories.

dec and was recrystd from MeOH, mp 327° dec, $\lceil \alpha \rceil^p$ -67° $(c1, \text{MeOH})$. Anal. $(C_{20}H_{28}N_2O_2 \cdot 0.5\dot{H}_2O)$ C, H, N.

 17β -(N³-Acetureido)androst-5-en-3 β -yl Acetate (38).—A stirred mixt of 280 g of 9 and 4 l. of Ac₂O was maintained at 50° for 48 hr. It was then poured into 15 l. of a $50:50$ ice-H₂O mixt contg 20 ml of Pyr and was stirred until all of the Ac20 had been destroyed. The cryst ppt was sepd by filtration and washed with H_2O to remove most of the HOAc. The ppt was then dissolved in CHCl₃ (1000 ml) and washed sequentially with H_2O (100 ml), satd ag NaHCO₃ until the washings were basic, and again with H_2O (100 ml). The dried soln was evapd to dryness under reduced pressure. The product 38 was recrystd from MeOH, yield 245 g, 69.8% , mp 210° , resolidifies and remelts at $227-230^{\circ}$, $[\alpha]_{\text{D}} -18^{\circ}$. *Anal.* $(C_{24}H_{36}N_2O_4)$ C, H, N.

 17β -(N³-Acetureido)androst-5-en-3 β -ol (39).—To a stirred soln of 38 (245 g) in anhyd MeOH under N_2 at room temp was added NaOMe (33 g). The mixt was stirred under N_2 for 16 hr and was then quenched by the addn of gl HOAc (40 ml). The reaction mixt was coned to about 1 l. under reduced pressure and dild with 4 1. of H_2O . The cryst product 39 was sepd by filtration and washed with H_2O (200 ml). The wet filter cake was dissolved in CHCl₃ (500 ml), washed with H₂O (2 \times 50 ml), dried, and coned *in vacuo.* The residue was recrystd from MeOH, yield 75.8%, mp 221-224°, [a]D -31.7° (Anal. (C₂₂- $H_{34}N_{2}O_{3}$ C, H, N.

 17β -Ureidoandrosta-1,4-dien-3-one (8).—A mixt of 40 (87 g), MeOH (3200 ml), and 2.5 *N* HC1 (800 ml) was heated at reflux for 1 hr. The solvents were removed *in vacuo* in a rotating evaporator at 50°, leaving an oily emulsion which was extd with CHCl₃ (3 \times 200 ml). The CHCl₃ exts were washed with H₂O to neutrality after which the solvent was distd *in vacuo*. The to neutrality after which the solvent was distd in vacuo. residue 8 was crystd from Me₂CO and had mp 248-249°, [a]D $+47.6^{\circ}$, yield 51.8 g. Anal. (C₂₀H₂₈N₂O₂)C, H, N.

Using this same procedure 11 was converted to 17β -uriedoandrost-4-en-3-one (7), mp 208-209°, from MeOH or EtOAc, $[\alpha]$ D+74.9. *Anal* (C₂₀H₂₀N₂O₂) C, H, N.

Testosterone Carbamate (3).—The reagents and conditions used are those of Loev and Kormendy.²⁴ To a stirred suspension of testosterone (1 g), NaOCN (455 mg), and CH_2Cl_2 (10 ml) at room temp was added $(F₃CCO)₂O$ (0.52 ml) in $CH₂Cl₂$ (1 ml) over 10 min. The resulting mixt was stirred overnight, and then dild to 50 ml with CH_2Cl_2 . It was washed with H_2O (15 ml). The dried CH₂Cl₂ soln was evapd in vacuo at <50° leaving a residue which afforded 700 mg of cryst 3, from MeOH. The anal, sample was recrystd successively from MeOH and EtOAc and showed mp 158-160°, $[\alpha]_D$ +109.9°, (c 1, MeOH). Anal. $(C_{20}H_{29}NO_3) C, H, N.$

17 β -Propionamidoandrost-4-en-3-one (6).—To a cooled (0°) soln of 17 β -aminoandrost-5-en-3-one cyclic ethylene acetal⁷ (331) mg) in Pyr (5 ml) was added n-PrCl (0.25 ml) . The mixt was stirred at room temp overnight. $H₂O$ was added to the homogeneous reaction mixt and the ppt which formed was sepd by filtration, washed well with H_2O , and dried. Two recrystns from MeOH afforded the anal. sample of 17β -propionamidoandrost-5-en-3-one cyclic ethylene acetal (42). *Anal* (C24- $H_{37}NO_4$) C, H, N. The above dioxolane (250 mg) was dissolved in MeOH (7 ml) and treated with p -TsOH (300 mg) and H₂O (1 ml). The stirred soln was maintained at room temp overnight and then dild with H_2O until no further pptn occurred. The product was collected by filtration, washed with H_2O , and dried, yield 160 mg. Crude 6 was recrystd from MeOH-EtOAc for anal., mp 230-231°, $[\alpha]_D + 55^\circ$. *Anal.* $(C_{22}H_{33}NO_2)$ C, H, N.

Compd 7 from 17β -Aminoandrost-5-en-3-one Cyclic Ethylene Acetal.—To a hot soln of the dioxolane (6.7 g) in EtOH (65 ml) was added a soln of KOCN $(9.5 g)$ in $H₂O$ $(13 ml)$. Heating was contd until a clear soln resulted. Aq 2.5 *N* HC1 (40 ml) was then added to the hot soln at such a rate that a $pH > 7$ was maintained. The mixt was heated at reflux for 15 min after addn of the acid. A ppt formed. For cooling to room temp H_2O was added to ppt the product, 17 β -ureidoandrost-5-en-3-one cyclic ethylene ketal, which was washed well with H_2O , dried, and recrystd from MeOH to which a drop of Pyr had been added, mp 320° dec. Anal. $(C_{22}H_{34}N_2O_3)$ C, H, N. This compd was converted to 7 by the method used in the conversion of 42 to 6.

17 β -Amino-3-methoxy-1,3,5(10)-estratriene (43).—A soln of 3 -methoxyestra-1,3,5(10)-trien-17-one oxime¹⁰ (9.0 g) in EtOH (1600 ml) was heated to boiling. Pieces of Na (total 185 g) were added until solids began to separate. EtOH (200 ml) was added to dissolve the solid. The mixt was poured into H_2O (11.) and the solvents were distd *in vacuo* to a dry residue. The solids were suspended in H₂O and extd with Et₂O $(3 \times 200 \text{ ml})$. The Et₂O exts were washed with H_2O , dried, and evapd leaving a residue which crystd from Et_2O . The product 43, 5.8 g, had mp 110-120° raised to 119-121° by recrystn from CH_2Cl_2 hexane. $Anal.$ (C₁₉H₂₇NO) C, H, N.

 17β -Acetamido-3-methoxy-1,3,5(10)-estratriene (28).—A soln of 43 (250 mg) was acetylated with $Pyr-Ac₂O$. After the usual work-up there was obtained 28 (260 mg), mp 243-244° (from MeOH), $[\alpha]_D$ -26°. Anal. (C₂₁H₂₉NO₂) C, H, N.

17 β -Acetamidoestr-4-en-3-one (20).—A sample of 17 β -amino-3-methoxy-1,3,5(10)-estratriene (43) (2.0 g) was reduced with Li-NH₃ (standard Birch reducing conditions) to 17β -amino-3methoxy-estra-2,5(10)-diene, which was not purified. The crude diene was acetylated with $Pyr-Ac₂O$ by the usual methods, worked up in an unexceptional manner, and then treated in aq MeOH with 2.5 *N* HC1 on the steam bath for 15 min. The reaction mixt was dild with H_2O and extrd with $CHCl₃$. The ext was washed, dried, and evapd leaving an oil which was chromatographed on alumina, the product being eluted with $Et₂O CHCI₃$ mixts. The combined eluates were recrystd from EtOAc and then from MeOH yielding 20, mp $236-238^\circ$, $[\alpha]_D$ -6.1°. *Anal.* $(C_{21}H_{31}NO_2)$ C, H, N.

17 β -Acetamidoandrost-4-en-3 β -ol (13).—To a soln of 5 (0.5 g) in 20 ml of THF was added 1.0 g of $LiAl(O-tert-Bu)_{3}H$. After stirring at room temp for 48 hr the reaction mixt was poured into ice-H₂O satd with NaH_2PO_4 . The product was extd into CHCl₃ which was washed with H_2O , dried, and evapd. The residue was crystd twice from CHCl₃ affording 13, mp 236-238°, α ₁ + 11.4°. *Anal.* $(C_{21}H_{33}NO_2) C, H, N$.

 17β -Acetamidoandrosta-4,6-dien-3-one (24).—Using the chloranil procedure essentially as described by Patchett, et al. [footnote *g,* Table I], 5 was converted to 24, mp 250-252°, from EtOAc, $[\alpha]$ D -26.2°. *Anal*. $(C_{21}H_{29}NO_2) C, H, N$.

17 β -Acetamido-5 α -androstan-3-one (15). To a suspension of 17β -acetamido-5 α -androstan-3 β -ol^{6b} (300 mg) in 15 ml of Me₂CO, cooled and stirred in an ice-bath, was added dropwise Jones reagent (0.67 ml) . After 10 min H₂O was added until no more pptn occurred. The product was extd with CHCl₃, washed with $H₂O$, dried, and evapd leaving an oil which was chromatographed on alumina. Et₂O eluted the product (130 mg) 15, mp 228-230°, from EtOAc.

3 β -Hydroxyandrost-5-en-17 β -yl Guanidine Hydroacetate Monohydrate (12) . - A soln of methylisothiourea \cdot HCl (6.7 g) in 30 ml of $H₂O$ was added to a stirred suspension of 34 (10.0 g) in 200 ml of A~-methylpyrrolidinone. The stirred mixt was treated with KOAc (12 g) and heated to reflux which effected dissoln of the components and was accompanied by evoln of CH?SH. After 6 hr at reflux, the cooled reaction mixt was poured into cold H_2O . The ppt which formed was sepd by filtration, washed successively with H_2O , MeOH, and Et₂O, and dried *in vacuo*. The crude product weighed 8.5 g and was recrystd from gl HOAc affording 7.17 g of 12, mp $>300^{\circ}$. This product gave a positive Sakaguchi test for guanidines. Repeated elemental analyses failed to give satisfactory values for this prepn. In order to obtain a satisfactory analysis, 12 was converted to $176-(N^2,N^3$ diacetylguanidino)androst-5-en-3 β -yl acetate (45) as follows. A suspension of 12 (250 mg) in DMF (5 ml) was treated with Et₃N (2.5 ml) and Ac₂O (1.5 ml) and the mixt was heated in a steam bath until the steroid dissolved *(ca.* 15 min). The cooled soln was dild with ice-H₂O, and, after the excess Ac₂O had reacted, was extd with EtOAc $(3 \times 10 \text{ ml})$. The combined exts were washed successively with H₂O, 10% aq NaHCO₃, and satd aq NaCl. The solvents from the dried exts were removed *in vacuo* leaving a residue which was adsorbed from C_6H_6 on alumina. The product 45 (90 mg) was eluted with C_6H_6 and crystd from EtOAc, mp 177-178°. *Anal* (C26H39N302) C, H, N.

 17α -Ureidoandrost-5-en-3 β -ol (10) was prepd from 17α aminoandrost-5-en-3 β -ol by the method described above for the prepn of 9, mp $>300^{\circ}$ dec, $[\alpha]_{D}$ -72° (c 0.5, MeOH). *Anal.* $(C_{20}H_{32}N_2O_2) \tilde{C}$, H, N.

⁽²⁴⁾ B. Loev and M. F. Kormendy, *J. Org. Chem.,* 28, 4322 (1963).